

(Promoted by Govt. of Gujarat)

Regd. Office & Works: P.O. Petrochemicals – 391 346, Dist. Vadodara (Guj.) INDIA **Phone**.: + 91-265-2232681, 3061200, 6540463 **Fax**: +91-265-2232130, 2230031

Website: www.gacl.com

MATERIAL SAFETY DATA SHEET

PHOSPHORIC ACID

1. Identification of substance

Synonyms: Ortho-phosphoric acid; Mono Phosphoric Acid

CAS No. : 7664-38-2 **Molecular Weight** : 98.00

Chemical Formula : H3PO4 in H2O

Manufacturer/Supplier: GUJARAT ALKALIES AND CHEMICAS LTD.

PO: Dahej-392 130, Ta: Vagara, Dist. Bharuch, State Gujarat, INDIA

24 EMERGENCY TELEPHONE: CCR (02641 256133) (+919879604102)

2. Composition/data on Components

 Ingredient
 CAS No
 Percent
 Hazardous

 Phosphoric Acid
 7664-38-2
 86 %
 Yes

 Water
 7732-18-5
 14 %
 No

3. Hazards Identification

Emergency Overview

DANGER! CORROSIVE. CAUSES SEVERE IRRITATION AND BURNS TO EVERY AREA OF CONTACT. HARMFUL IF SWALLOWED OR INHALED.

Health Rating: 2 - Moderate Flammability Rating: 0 - None Reactivity Rating: 2 - Moderate Contact Rating: 3 - Severe (Corrosive)

Protective Equip: GOGGLES & SHIELD; PVC COAT & APRON; VENT HOOD; PROPER GLOVES

Storage Color Code: White (Corrosive)

Potential Health Effects

Inhalation:

Inhalation is not an expected hazard unless misted or heated to high temperatures. Mist or vapor inhalation can cause irritation to the nose, throat, and upper respiratory tract. Severe exposures can lead to a chemical pneumonitis.

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Ingestion:

Corrosive. May cause sore throat, abdominal pain, nausea, and severe burns of the mouth, throat, and stomach. Severe exposures can lead to shock, circulatory collapse, and death.

Skin Contact:

Corrosive. May cause redness, pain, and severe skin burns.

Eye Contact:

Corrosive. May cause redness, pain, blurred vision, eye burns, and permanent eye damage.

Chronic Exposure:

No information found.

Aggravation of Pre-existing Conditions:

Persons with pre-existing skin disorders or eye problems, or impaired respiratory function may be more susceptible to the effects of the substance.

4. First Aid Measures

Inhalation:

Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Call a physician immediately.

Ingestion:

If swallowed, DO NOT INDUCE VOMITING. Give large quantities of water. Never give anything by mouth to an unconscious person. Get medical attention immediately.

Skin Contact:

Immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Call a physician, immediately. Wash clothing before reuse.

Eye Contact:

Immediately flush eyes with gentle but large stream of water for at least 15 minutes, lifting lower and upper eyelids occasionally. Call a physician immediately.

5. Fire Fighting Measures

Fire

Not considered to be a fire hazard. Contact with most metals causes formation of flammable and explosive hydrogen gas.

Explosion:

Not considered to be an explosion hazard.

Fire Extinguishing Media:

Use any means suitable for extinguishing surrounding fire. Water spray may be used to keep fire-exposed containers cool. If water is used, use in abundance to control heat and acid build-up.

Special Information:

In the event of a fire, wear full protective clothing and NIOSH-approved self-contained breathing apparatus with full-face piece operated in the pressure demand or other positive pressure mode.

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6. <u>Accidental Release Measures</u>

Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Isolate hazard area. Keep unnecessary and unprotected personnel from entering. Contain and recover liquid when possible. Neutralize with alkaline material (soda ash, lime), then absorb with an inert material (e. g., vermiculite, dry sand, earth), and place in a chemical waste container. Do not use combustible materials, such as saw dust. Do not flush to sewer! Require reporting spills and releases to soil, water and air in excess of reportable quantities.

7. <u>Handling and Storage</u>

Keep in a tightly closed container. Protect from physical damage. Store in a cool, dry, ventilated area away from sources of heat, moisture, incompatibilities, and direct sunlight. Corrosive to mild steel. Store in rubber lined or 316 stainless steel designed for phosphoric acid. Do not wash out container and use it for other purposes. When diluting, the acid should always be added slowly to water and in small amounts. Never use hot water and never add water to the acid. Water added to acid can cause uncontrolled boiling and splashing. Protect from freezing. Containers of this material may be hazardous when empty since they retain product residues (vapors, liquid); observe all warnings and precautions listed for the product.

8. <u>Exposure Controls/Personal Protection</u>

Airborne Exposure Limits:

-OSHA Permissible Exposure Limit (PEL): 1 mg/m3 (TWA)

-ACGIH Threshold Limit Value (TLV): 1 mg/m3 (TWA), 3 mg/m3 (STEL)

Ventilation System:

A system of local and/or general exhaust is recommended to keep employee exposures below the Airborne Exposure Limits. Local exhaust ventilation is generally preferred because it can control the emissions of the contaminant at its source, preventing dispersion of it into the general work area. Please refer to the ACGIH document, Recommended Practices, most recent edition, for details.

Personal Respirators (NIOSH Approved):

If the exposure limit is exceeded, a full face piece respirator with high efficiency dust/mist filter may be worn up to 50 times the exposure limit or the maximum use concentration specified by the appropriate regulatory agency or respirator supplier, whichever is lowest. For emergencies or instances where the exposure levels are not known, use a full-face piece positive-pressure, air-supplied respirator. WARNING: Air purifying respirators do not protect workers in oxygen-deficient atmospheres.



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Skin Protection:

Wear impervious protective clothing, including boots, gloves, lab coat, apron or coveralls, as appropriate, to prevent skin contact.

Eye Protection:

Use chemical safety goggles and/or a full face shield where splashing is possible. Maintain eye wash fountain and quick-drench facilities in work area.

9. <u>Physical and Chemical Properties</u>

Appearance:

Clear, colorless syrupy liquid.

Odor: Odorless.

Solubility:

Miscible in all proportions in water.

Specific Gravity: (water = 1)

1.892 at 25 °C

pH:

1.5 (0.1 N agueous solution)

Boiling Point:

300 °C

Melting Point:

42.4 °C

Vapor Density (Air=1):

Not available

Vapor Pressure (mm Hg):

0.286 mm Hg at 20 °C

Evaporation Rate (BuAc=1):

No information found.

10. Stability and Reactivity

Stability:

Stable under ordinary conditions of use and storage. Substance can super cool without crystallizing.

Hazardous Decomposition Products:

Phosphorus oxides may form when heated to decomposition.

Hazardous Polymerization:

Will not occur.



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Incompatibilities:

Liberates explosive hydrogen gas when reacting with chlorides and stainless steel. Can react violently with sodium tetrahydroborate. Exothermic reactions with aldehydes, amines, amides, alcohols and glycols, azocompounds, carbamates, esters, caustics, phenols and cresols, ketones, organophosphates, epoxides, explosives, combustible materials, unsaturated halides, and organic peroxides. Phosphoric acid forms flammable gases with sulfides, mercaptans, cyanides and aldehydes. It also forms toxic fumes with cyanides, sulfide, fluorides, organic peroxides, and halogenated organics. Mixtures with nitromethane are explosive.

Conditions to Avoid:

Incompatibles.

11. Toxicological Information

Oral rat LD50: 1530 mg/kg; investigated as a mutagen.

Ingredient	Known	Anticipated	Category	
Phosphoric Acid (7664-38-2)		No	No	None
Water (7732-18-5)		No	No	None

12. Ecological Information

Environmental Fate:

When released into the soil, this material may leach into groundwater. When released to water, acidity may be readily reduced by natural water hardness minerals. The phosphate, however, may persist indefinitely. **Environmental Toxicity:**

No information found.

13. <u>Disposal Considerations</u>

Whatever cannot be saved for recovery or recycling should be handled as hazardous waste and sent to an approved incinerator or disposed in the approved waste facility. Processing, use or contamination of this product may change the waste management options. State and local disposal regulations may differ disposal regulations. Dispose of container and unused contents in accordance with state and local requirements.



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14. Transport Information

Domestic (Land)

Proper Shipping Name: PHOSPHORIC ACID

Hazard Class: 8 UN/NA: UN1805 Packing Group: III

International (Water.)

Proper Shipping Name: PHOSPHORIC ACID, LIQUID

Hazard Class: 8 UN/NA: UN1805 Packing Group: III

15. Regulatory Information

\Chemical Inventory Status - Part 1\							
Ingredient Phosphoric Acid (7664-38-2) Water (7732-18-5)	Yes	="	EC Yes Yes	Japan Yes Yes	<u>Australia</u> Yes Yes		
\Chemical Inventory Status - Part 2\							
Ingredient Phosphoric Acid (7664-38-2) Water (7732-18-5)	Canac Yes Yes	Korea_	DSL Yes Yes	NDSL No No	Phil. Yes Yes		
\Federal, State & International Regulations - Part 1\							
Ingredient Phosphoric Acid (7664-38-2) Water (7732-18-5)	RQNo	-SARA 302- TPQ No	<u>List</u> No No	Chemica No No	RA 313 <u>Il Catg.</u> No No		
\Federal, State & International Regulations - Part 2\							
Ingredient	CERCLA	-RCR	A- <u>261.33</u>	<u>8(d)</u>	<u>-TSCA-</u>		
Phosphoric Acid (7664-38-2) Water (7732-18-5)		5000 No	No No		No No		
Chemical Weapons Convention: No TSCA 12(b): No CDTA: No SARA 311/312: Acute: Yes Chronic: No Fire: No Pressure: No Reactivity: No (Pure / Liquid)							



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16. Other Information

NFPA Ratings: Label Hazard Warning: a) HAZARD:		Health: 2	Flammability: 0	Reactivity: 0		
		Cause	s irritation on contact w	rith skin/eyes. Corrosive to metals.		
b)	SAFETY:	Use R	ubber/PVC hand glove	s, safety goggles and face shield.		
c)	FIRST AID:		affected parts/eyes with es. Seek Doctor's advices.	n plenty of water for atleast 15 ce.		
e)	HANDLING:		Handle with care, do not drop the carboy. Place it on plain surface, free of stones, nails etc.			
f)	EMERGENCY S			ith plenty of water. Neutralise with lime/soda ash.		

Disclaimer:

Information contained in this material data sheet is believed to be reliable but no representation, guarantee or warranties of any kind are made as to its accuracy, suitability for a particular application or results to be obtained from them. It is upto the manufacturer/seller to ensure that the information contained in the material safety datasheet is relevant to the product manufactured / handled or sold by him as the case may be. The company makes no warranties, expressed or implied, in respect of the adequacy of this document for any particular purpose.



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